On the quantum statistics of bound states within the Rutherford model of matter

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Received 15 December 2011, revised 20 February 2012, accepted 7 March 2012 Published online 16 April 2012

The quantum statistical treatment of the Rutherford model is analyzed, considering matter as a system of point charges (electrons and nuclei). First, the solutions of different fundamental problems are discussed, such as the divergence of the partition function, elaborated by Herzfeld, Planck, Brillouin and Rompe. Beyond this analysis in the historical context, the modern state of art is presented and new results are given which explain why bound states according to a discrete part of the spectra occur only in a valley in the temperature-density plane. Based on the actual state of the quantum statistics of Coulomb systems, virial expansions within the canonical ensemble and the grand ensemble and combinations are derived. Furthermore, the transitions along isotherms are studied: (i) the formation of bound states occurring by increasing the density from low to moderate values, (ii) the disappearance of bound state effects at higher densities. Within the physical picture isotherms of pressure for hydrogen are calculated in a broad density region. It is shown that in the region between 20000 K and 100000 K and particle densities below 10^{22} cm⁻³ the cross-over from full to partial ionization may be well described by the contributions of extended ring diagrams and ladder diagrams.

having a positive central charge +Ne, and surrounded by a compensating charge of N electrons [1]. In Rutherford's model the atom is nearly empty, it is made up of a central charge and electrons and the glue keeping the system together are the Coulomb forces. Today we denote the positive central charge as the atomic nucleus, though Rutherford did not use the term "nucleus" in his paper. The central point charge is surround by a cloud of point electrons orbiting around the nucleus. The first quantummechanical treatment of the Rutherford model was given in 1913 by Bohr. According to the Bohr theory the energies and radii of the orbits are (in Gaussian units)

$$E_{s} = -\frac{\mu e^{4}}{2\hbar^{2}s^{2}}; \quad a_{s} = s^{2}a_{B};$$

$$a_{B} = \frac{\hbar^{2}}{\mu e^{2}}; \quad \mu = \frac{m_{e}m_{+}}{m_{e} + m_{+}}.$$
(1)

Here and in the following we denote the so-called main quantum number by *s* in order not to be mixed with the density *n*. According to the Bohr model there are infinitely many levels close to the series limit $s \to \infty$. A first statistical theory of the Bohr model was developed already in 1913 by the Austrian physicist Karl Herzfeld [2]. Herzfeld detected that the Bohr model had a serious de-

1 Introduction

In May 1911, Rutherford came forth with a model for the structure of atoms which provided a first understanding why electron scattering can so deeply penetrate into the interior of atoms, so far unexpected experimental results [1]. Rutherford explained his scattering results as the passage of a high speed electron through an atom

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Original Paper

ficiency, the internal atomic partition function

$$\sigma(T) = \sum_{s,l,m} \exp\left(-\beta E_s\right) = \sum_{s=1}^{s_m} s^2 \exp\left(\frac{I}{s^2 k_B T}\right);$$

$$I = \frac{\mu e^4}{2\hbar^2 s^2}.$$
(2)

is divergent ($I = -E_1$: ionization energy).

We see that the contributions to the internal atomic partition function for hydrogen according to the definition by Eq. (2) increase as s^2 (see Fig. 1) and the sum can get any value in dependence on the maximal number s_m

and is divergent. This problem exists for all elements. It is inherent in any Coulomb system, e.g. the electron - hole plasma in excited semiconductors that will not be considered here. Here we deal mainly with hydrogen which is the most abundant element in the Universe, its physical properties raised, in the past, many works.

The development of the quantum statistics of plasmas is intimately connected with the requirements of a new scientific discipline - astrophysics. In order to bring some systematics into the existing observations, to understand why spectra appeared under definite conditions, a theory was required which provides the abundan-

Figure 1 (online color at: www.ann-phys.org) Left panel: The terms in the atomic partition function defined by Eq. (2) for two temperatures corresponding to $\beta I = 5$ (red and green curves above) and $\beta I = 10$ (blue and purple curves below). Up to terms of order $E_{s_m} \simeq k_B T$ the terms in the usual partition function decrease monotonically and then they start to increase as s^2 lead-

ing to the Herzfeld divergence problem. For comparison we show the Brillouin-Planck-Larkin renormalized partition function with the terms of a convergent series. Right panel: The region in the density-temperature plane of hydrogen where the formation of bound states is expected, see text.

cies of atomic energy levels in stars. Such a theory was created on the basis of the Rutherford-Bohr model by the work of Eggert and Saha around 1920. Their work is an application of thermodynamics to the Rutherford-Bohr model, which was inspired by the papers of the Nernst school. The ionization problem was studied within the chemical picture and using Sackur's and Tetrode's statistical expression for the entropy of gases. Eggert published a first approach in 1919 [3], his paper came to the attention of the young bengalic physicist Saha, who succeeded to improve the approach considerably [4,5]. Saha considers, in the paper from 1921, Calcium plasmas as an example. Formulating his equation for hydrogen we get for the fraction of ionized atoms α the relations

$$\frac{1-\alpha}{\alpha^2} = nK; \quad K = \Lambda^3 \exp(\beta I); \quad \Lambda = \frac{h}{\sqrt{2\pi\mu k_B T}}.$$
 (3)

The solution of this quadratic equations is given by

$$\alpha = A(\gamma), \quad \gamma = nK(T), \quad A(x) = \frac{1}{2x} \left[\sqrt{1+4x} - 1 \right]. \tag{4}$$

Considering an ideal mixture of electrons, ions and atoms we obtain for the pressure

$$\beta p = (1 + A(\gamma))n. \tag{5}$$

We call the function $A(\gamma)$, which monotonically decreases with the density, the Saha function.

At high densities where the mean distance of protons approaches the Bohr radius, this leads to an enormous pressure acting on the neutrals which will finally be destroyed; we note however that this limit is not described in the present approach. The region were bound states may exist is shown in Fig. 1.

Eggert and Saha used the thermodynamic approach of Sackur and Tetrode which needs only the binding energy, avoiding this way the problem of divergence of the partition function. For more accurate treatment of astrophysical tasks the most important problem was the divergence of the partition function of the Bohr atom. As already said above, the terms in the atomic partition function for hydrogen according to the definition by Eq. (2) diverge as s^2 (see Fig. 1). The easiest way is to cut the sum at the lowest terms corresponding to $|E_s| \simeq k_B T$. This way of solving the divergence problem is due to the classical paper of Planck published in 1924 [6], later in 1938 applied to special plasma problems by Riewe and Rompe [7]. Developing Plancks approach Brillouin proposed in 1932 a more smooth way of removing the divergence which led to the formula

$$\sigma_{\rm BPL}(T) = \sum_{s=1}^{\infty} s^2 [\exp(\beta I/s^2) - 1 - (\beta I/s^2)].$$
(6)

This procedure was justified only in 1960 by Larkin [8] and subsequently by the present authors in collaboration with D. Kremp by using strict quantum-statistical methods [9–14]. For later purpose it is interesting to note, that this so-called Brillouin-Planck-Larkin (BPL) partition function may be expressed (just after developing the exponential function) by a different series which contains the Riemann ζ function

$$\sigma(T) = \sum_{k=2}^{\infty} \frac{\zeta(2k-2)(\beta I)^k}{k!}.$$
(7)

In the present paper we will give a systematic quantumstatistical statistical theory of matter within the Rutherford model, including in particular the equation of state. We concentrate on the so-called physical picture which stays within the Rutherford picture and avoids the introduction of chemical species like atoms, molecules etc. We note that in some earlier work the Saha theory has been proven to become asymptotically exact in zerotemperature and zero-density limit [15-17]. Here we restrict ourselves mainly to a region of practical interest, i.e., to temperatures between 20000 K and 100000 K and particle densities between 10^{18} and 10^{22} per cm³. In some of our earlier work the transition to a chemical picture was made following the principle of equivalence that bound states are to be treated on the same footing as free particles [10, 13, 18]. This approach was guite successful in the description of partial ionization [13, 19–23]. On the other hand, one has to accept that in the region of high densities several serious difficulties appear which are mainly connected with the impossibility of a clear distinction between free and bound states in a dense system [24]. This is the reason why we stay here within the physical picture taking into account the important higher order terms [25]. In a similar spirit, but using different techniques, a consistent treatment of bound state contributions within the physical picture has been constructed already by Alastuey et al. [26-28]. We show here first a treatment of the Coulomb singularities in the framework of the virial expansions within the canonical ensemble, then we discuss the fugacity expansions and identify the higher order contributions which allow a Saha-type approximation. In the last part we investigate the role of the identity of electrons within the grand canonical ensemble.

In the first part of this work we will to show that the Brillouin-Planck-Larkin procedure provides the most natural way to avoid the divergencies for non-degenerate systems. In another work [41] we study degenerate systems and show that a consequent treatment of the identity of electrons in the Hartree-Fock approximation including bound states is a key for the treatment of hydrogenic bound states. The present approach is based on the method of Green's functions as developed in [13,14,18,29, 30].

Any successful description of plasmas has to go beyond the Saha theory which is something like the lowest density order approximation for plasmas. This way our strategy should be to obtain first an approximation equivalent to the Saha description and then to go further. The derivation of Saha-type approximations from quantum statistics was first studied by Planck, Fowler and Brillouin [6, 31, 32]. Since then it is a topic of permanent interest due to the importance of Saha-type equations for experimental work and for many technological applications. A first big progress in the strict quantum statistical treatment was achieved in the 60th by papers of Montroll, Ward, Vedenov, Larkin, Abrikosov and others [8,33,34] and in subsequent work [9,10,12–14,18,35–38]. We note that a consistent description of bound states in macroscopic physical systems is a hot problem of modern quantum statistics and was treated also independently and using different techniques by other workers [26-28].

2 The model of Planck for hydrogen plasmas

Planck's work was inspired by the earlier papers of Eggert and Saha that was an application of thermodynamics and Sackur's and Tetrode's expression for the entropy. Accordingly Eggert and Saha needed only the binding energy, avoiding this way the problem of divergence of the partition function. In his Annalen paper from 1924, Max Planck uses a more consequent statistical approach based on the partition function. He splits first the atomic partition function into three contributions

$$\sigma(T) = \sigma_3(T) + \sigma_2(T) + \sigma_1(T). \tag{8}$$

where $\sigma_3(T)$ is the contribution of the lowest bound states $|E_s| > k_B T$, $\sigma_2(T)$ is the contribution of the bound states below the series limit and $\sigma_1(T)$ is the entire remainder. Using a rather complicated quasiclassical derivation Planck was able to show that due to compensation effects it holds

$$\sigma_2(T) + \sigma_1(T) \simeq \frac{V}{\Lambda^3}.$$
(9)

In other words the sum of the second and the third contributions reduces to the partition function of an ideal gas, i.e. the all diverging contributions cancel each other. Planck's paper was considered in those years to be very important, so it was discussed in some detail in several

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textbooks and monographs, e.g. in the books written by Brillouin [32] and Fowler [31]. We follow here mostly Brillouin. According to Brillouin [32], the cancellation effect found by Planck is exactly true only if one makes the particular choice

$$\sigma_3(T) = \sigma_{\rm BPL}(T). \tag{10}$$

As the derivations of Planck, Fowler and Brillouin are very difficult to understand we give the proof of Planck's important statement in a more easy version which however uses the same physical assumptions [9]. We start with Planck's final expression

$$F = k_B T N[\alpha [\ln(n\Lambda_e^3 \alpha) + 1] + \alpha [\ln(n\Lambda_i^3 \alpha) + 1] + (1 - \alpha) [\ln(n\Lambda_a^3 (1 - \alpha)) - \ln(\sigma_3(T)) + 1]] + F_{\text{ex}}.$$
 (11)

Here F_{ex} is the excess part of the free energy. Planck and Brillouin claim that this contribution is zero, $F_{ex} = 0$; however these researchers did not include screening, an effect detected nearly at the same time by Debye. In the following part of this section we repeat the derivation of the excess part but include also the contributions from screening effects using a semiclassical screened cluster expansion approach [9, 39]. In order to find the excess free energy of a plasma we have to calculate all contributions beyond the ideal contributions of electrons, protons and atoms. Following [9, 10] and using the classical technique of screened virial expansions [39] we find including the Debye limiting law and the second virial coefficents

$$F_{ex} = -k_B T V \left[\frac{\kappa^3}{12\pi} + n_e^2 \tilde{B}_{ee} + 2n_e n_i \tilde{B}_{ei} + n_i^2 \tilde{B}_{ii} \right] + \dots,$$
(12)
$$\tilde{B}_{ab} = 2\pi \int dr \, r^2 \left[S_{ab} \exp(g_{ab} + \beta V_{ab}) - 1 - g_{ab} - \frac{1}{2} g_{ab}^2 \right].$$
(13)

The Boltzmann factors without bound state contributions are given by [9]

$$S_{ee}(r) = S_{ii} = \exp(-\beta e^2/r), \quad V_{ab} = e_a e_b/r,$$

$$g_{ab} = V_{ab} \exp(-\kappa r)$$
(14)

$$S_{ie}(r) = \exp\left(\frac{\beta e^2}{r}\right) - \frac{4\pi}{\sqrt{2\pi\mu k_B T}} \int_0^{p_0} dp \, p^2$$
$$\times \left[\exp(-\beta E(p,r)) - 1 + \beta E(p,r)\right], \tag{15}$$

$$E(p,r) = \frac{p^2}{2\mu} - \frac{e^2}{r}; \quad p_0 = \sqrt{\frac{2\mu e^2}{r}}; \quad \kappa^2 = 8\pi n\beta e^2.$$
(16)

Here the second contribution in S_{ie} is just subtracting the bound state contribution from the full Boltzmann factor in a classical approximation. Up to the Debye contribution, which was not yet known to Planck, since is was just in print when Planck wrote his paper, this is just what Planck and Brillouin calculated. After tedious calculations one can show analytically with a numerical check [9], that the sum of the integrals beyond the Debye term cancel each other and gives zero.

$$F_{\rm ex} = -k_B T V \left[\frac{\kappa^3}{12\pi} + \mathcal{O}(n^{5/2}) \right]. \tag{17}$$

The excess free energy consists in the given quasiclassical approximation only of the Debye term and some higher order corrections. In particular, there is no contribution in the quadratic order. This way we could justify the remarkable result of Planck and Brillouin, we just found a correction, the Debye term. Including this essentially classical term we get

$$\beta p = (1+\alpha)n - \frac{1}{2}(8\pi\beta e^2)^{3/2}(\alpha n)^{3/2},$$
(18)

 $\alpha = A(\gamma)$ according Eq. (4). A typical picture of the pressure in the Saha approach shows that bound states exist only in certain valley in the density-temperature plane, see Fig. 1. We will see later that the bound state contributions as well as the Debye contributions are responsible for a reduction of the pressure in comparison to the ideal pressure (see Fig. 2). The Saha equation and the corresponding pressure do not describe the region of high densities, the Debye term yields just a first correction. At higher densities another important influence is due to the Fermi pressure in the region of electron degeneracy. In order to check for the influence of electron degeneracy we compare with the ideal gas pressure including the Fermi pressure of the electrons:

$$\beta p_e^F(\beta, n) = n[1 + 0.08839n\Lambda_e^3 - 0.00083n^2\Lambda_e^6 + 0.000012n^3\Lambda_e^9 + \cdots];$$

$$\Lambda_e = \frac{h}{\sqrt{2\pi m_e k_B T}}.$$
(19)

Looking at typical isotherms (see e.g. Figs. 2, 3) we see that the pressure related to the ideal pressure decreases with growing density and the pressure increases again. This demonstrates that the region of bound state formation where the pressure is significantly smaller than the ideal pressure of an electron-proton gas is a limited region in the density-temperature plane (the valley of bound states).

The main task of this work is the study of the two transitions:

(i) the formation of bound states occurring at the crossover from fully ionization to the region were bound states are formed (we have partial ionization) observed with increasing density,

(ii) the destruction of bound states beyond the valley of bound states in the high density region.

The region of our interest in the density-temperature plane where bound states appear is demonstrated in Fig. 1. We determine here the first transition by the densities where $p/2nk_BT$ decreases below the value 3/4and identify the second transition by the condition that $p^{F}/2nk_{B}T$ exceeds the value 1, that means the Fermi pressure p^F gives the overall dominant contribution. From the physical point of view we should expect that the real pressure should be nearer to the highest contribution, i.e at lower densities nearer to the pressure given by the Saha equation and at high densities nearer to the Fermi pressure. Later we will confirm the result that bound states occur only on an island in the T - nplane. In our primitive extended Saha model, the formation of atoms is described by the mass action law which is part of the Saha model and the destruction at high density is described in a very rough approximation only by the strong increase of the ideal Fermi pressure. In the following sections we will show how these effects are more correctly described by methods of quantum statistics. We present in Fig. 1 first a horizontal line corresponding to the density of $n = 10^{23}$ cm⁻³. For such a density the the mean distance between protons (or electrons) defined by

$$\frac{4\pi}{3}nr_0^3 = 1$$
 (20)

is around the Bohr diameter.

$$r_0 \simeq 2a_B; \quad na_B^3 \simeq 0.03.$$
 (21)

For larger densities (smaller average distances) the formation of bound states may be excluded, the plasma behaves nearly as an ideal Fermi gas since there is not enough space for forming atomic orbitals or molecular orbitals. The disappearance of bound states is referred to as Mott condition, corresponding to the fact that atoms are destroyed if the mean distance of the electrons is much smaller than the Bohr radius. Such situation was discussed in [30, 40, 41]. For more detailed information see [14, 36]. We show, in Fig. 1, a line $\alpha \simeq 0.5$. The Saha equation provides for the condition $\alpha = 0.5$ the estimate

$$\gamma_0 = nK(T) = n\Lambda^3 \exp(I/k_B T) = 2.$$
 (22)

We are here mainly interested in the region of atomic bound states. Near to the upper border of the corridor region shown in Fig. 1 only the atomic ground state exists, but near to the lower border many bound states are to be expected and we have to avoid the divergence of the partition function by appropriate renormalization. At lower temperatures $T \le 30\,000$ K we may expect also the formation of molecules in a certain density region.

3 Derivation of the Brillouin-Planck-Larkin partition function by low-density virial expansions

Following our earlier papers [12, 35] and books [13, 14] we use the method of cluster expansions in order to derive the exact virial functions. We start here with a treatment of the Coulomb contributions in the framework of the virial expansions within the canonical ensemble. For a real quantum gas with only short range forces the free energy may be at low density described by a virial expansion [12, 35]

$$F = F_{id}^B - k_B T V \left[\sum_{ab} n_a n_b B_{ab} + \sum_{abc} n_a n_b n_c B_{abc} + \cdots \right].$$
(23)

$$B_{ab} = \operatorname{const} \operatorname{Tr}[\exp(-\beta H_{ab}) - \exp(-\beta H_{ab}^{0})], \qquad (24)$$

where the Hamiltonian of pairs of species a and b is defined as

$$H_{ab} = -\frac{\hbar^2}{2\mu_{ab}}\Delta + V_{ab}.$$
(25)

Using the resolvent representation for the exponential operator we get

$$B_{ab} = \frac{4\pi^{3/2}(1+\delta_{ab})\lambda_{ab}^3}{(2s_a+1)(2s_b+1)} \frac{1}{2\pi i} \int_c \exp(-\beta z)F(z)dz,$$

$$\lambda_{ab} = \frac{\hbar}{\sqrt{2\mu_{ab}k_BT}},$$

$$F(z) = \operatorname{Tr}\left[\frac{1}{H_{ab}-z} - \frac{1}{H_{ab}^0-z}\right].$$
(26)

After some transformation we arrive at the following representation by Jost functions $D_{\ell}(z)$

$$F(z) = \frac{(2s_a + 1)(2s_b + 1)}{(1 + \delta_{ab})} \sum_{\ell=0}^{\infty} (2\ell + 1)$$
$$\times \left[1 \pm \delta_{ab} \frac{(-1)^{\ell}}{(2s_a + 1)} \right] (-1) \frac{d}{dz} \ln D_{\ell}(z).$$
(27)

The Jost functions $D_{\ell}(z)$ are analytical functions with poles at the bound states and a branch cut at the positive real axis defined by the scattering phase shifts. These

functions as well as other scattering quantities are exactly known for Coulomb systems. However in the procedure of calculating the integrals there is a difficulty due to the long range character of Coulomb forces and the necessity to introduce some screening procedure. In analogy to the classical case discussed in the previous section we may write including screening effects

$$F = F_{id}^{B} - k_{B}TV$$

$$\times \left[\frac{\kappa^{3}}{12\pi} + \sum_{ab} n_{a}n_{b}\tilde{B}_{ab} + \sum_{abc} n_{a}n_{b}n_{c}\tilde{B}_{abc} + \cdots\right].$$
(28)

Here the tilde denotes that screened potentials were included into the virial coefficients. In order to reduce the difficulties connected with a complete derivation we proceed as follows: The screening procedure is necessary only for the lowest orders of the interaction parameter e^2 . Higher orders in e^2 beginning with e^8 have already a decay rate of e^8/r^4 and do not need any screening. Taking into account that screening refers to the low orders we split the second virial coefficients into two parts

$$\tilde{B}_{ab} = G'\tilde{B}_{ab} + G''\tilde{B}_{ab}; \quad G''\tilde{B}_{ab} = G''B_{ab},$$
(29)

$$G'\tilde{B}_{ab} = O(1) + O(e^2) + O(e^4) + O(e^6),$$

$$G''B_{ab} = O(e^8) + O(e^{10}) + \cdots.$$
(30)

Let us first calculate the higher orders, remembering that according to Brillouin and Larkin, here the contribution of the bound states are to be expected. Introducing here the known Jost functions, the final result reads

$$G''B_{ab} = 2\pi^{3/2}\lambda_{ab}^{3}[1 + \beta e_{a}e_{b}\kappa] \sum_{m \ge 4} \frac{\xi_{ab}^{m}}{2^{m}\Gamma(m/2+1)} \times \left[\zeta(m-2) \pm \delta_{ab}\frac{(1-2^{2-m})}{2s_{a}+1}\zeta(m-1)\right].$$
(31)

The contributions in lower order have to be calculated term by term. The result of tedious calculations is (C = 0.577 – Euler's constant)

$$\begin{aligned} G'\tilde{B}_{ab} &= 2\pi\lambda_{ab}^{3}[1+\beta e_{a}e_{b}\kappa] \\ &\times \left[-\frac{1}{6}\xi_{ab} - \frac{\sqrt{\pi}}{8}\xi_{ab}^{2} - \frac{1}{6}\left(\frac{1}{2}C + \ln 3 - \frac{1}{2}\right)\xi_{ab}^{3} \right] \\ &+ \frac{\pi}{3}(\beta e_{a}e_{b})^{3}[(1+\beta e_{a}e_{b}\kappa)\ln(3\kappa\lambda_{ab} - \beta e_{a}e_{b}\kappa(1-\ln(4/3))] \\ &\pm \delta_{ab}[1+\beta e_{a}e_{b}\kappa] \left[\frac{\sqrt{\pi}}{4} + \frac{\xi_{ab}}{2} + \frac{\sqrt{\pi}(\ln 2)}{4}\xi_{ab}^{2} + \frac{\pi^{2}}{72}\xi_{ab}^{3} \right]. \end{aligned}$$
(32)

Note that most difficulties are connected here with the term of order ξ_{ab}^3 [13, 14, 36].

For charge and mass symmetrical systems (all masses are equal and the reduced masses are just half of it) the odd contributions cancel and we get in the case of hydrogen for the sum, taking into account $\beta I = \xi^2/4$:

$$\sum_{ab} G'' B_{ab} = 2\pi \sqrt{\pi} \sum_{k=2}^{\infty} \frac{\zeta (2k-2)\xi^{2k}}{2^{2k}k!} = 4\pi \sqrt{\pi} \lambda_{ie}^3 \sigma_{\rm BPL}(T).$$
(33)

This way we have given a strict derivation of the Brillouin-Planck-Larkin partition function given by Eq. (6). The final results for the equation of state of hydrogen may be expressed in form of a density expansion:

$$\beta p(\beta, n) = 2n - 2n(8\pi\beta e^2)^{3/2}\sqrt{n} - 2n^2[\Lambda^3\sigma_{\rm BPL}(T)] \\ \times \left[1 + \frac{3}{2}\beta e^2\kappa\right] - 2n^2K^*(T)] + L(T)n^{5/2} + \mathcal{O}(n^3\log n).$$
(34)

The function $K^*(T)$ which appears in the formula Eq. (34) depends only of the temperature. This transcendent function is exactly known [13, 14, 36] and may be calculated as an infinite series in ξ using the expressions for B_{ab} given above. There are also tables for hydrogen available [13]. However since the full expression is a quite long and not easy to handle, we will give here only a rather good (in numerical respect) and simple approximation

$$K^{*}(T) = -\frac{1}{2}\Lambda^{3}\beta I - \frac{1}{8\sqrt{2}}\Lambda^{3}.$$
(35)

This expression is exact with respect to the largest asymptotic term $\mathcal{O}(I/k_B T)$ and was checked with the available tables [13]. As a technical remark we note, that the terms not contained in the approximation Eq. (35) are mainly given by the contributions of the odd powers in ξ to the pressure. We do not give here a detailed discussion since these terms give only a small contribution to the pressure in the region 10^4 K < $T < 10^5$ K. The exact expression may be found in original articles [10, 12] and in several books [13, 14, 36]. The complete term of order $n^{5/2}$ has also been computed in several works [37, 42, 43]. We will not discuss this matter here since we use here some approximations for L(T) and because this function does not contain bound state contributions. There were different attempts to determine such contributions $n^{5/2}$ in the framework of Green's function techniques. Here we mention [37, 38]. These results should be compared to the corresponding expressions for the $n^{5/2}$ terms given by [43, 44]. Though there are results which seem to be in agreement with each other, there remains still some open

questions [41], see [45]. Here we are interested mainly in the bound state effects and restrict ourselves to the asymptotic results for the case $\xi_{ie}^2 \gg 1$ what is equivalent to $T \ll I/k_B$. For the case of hydrogen and lower temperatures $T \ll I/k_B$ we get in asymptotic approximation the final results for the pressure at small densities:

$$\beta p(\beta, n) = n \left[1 + \frac{1}{8\sqrt{2}} n \Lambda_e^3 + \cdots \right]$$
$$+ n - 2n \frac{1}{2} (8\pi \beta e^2)^{3/2} \sqrt{n} - n^2 \Lambda^3(\beta I) + n^2 \Lambda^3 \sigma_{\text{BPL}}(T)$$
$$\times \left[1 + \frac{3}{2} \beta e^2 \kappa \right] + \cdots + \mathcal{O}(n^{5/2}) + \mathcal{O}(n^3 \log n).$$
(36)

This way we have shown that in the quantum-statistical expressions for the thermodynamic functions appears the Brillouin-Planck-Larkin partition function in a natural way as a convergent infinite sum starting with terms of order e^8 . As mentioned in the previous section, a first approximation to this partition function was obtained by Planck [6] and a first explicit derivation was given by Brillouin [32]. Later an asymptotic quantum-statistical derivation was given by Larkin [8]. We follow here Larkin's asymptotic approach which was extended later in [10] but include more terms than in the earlier work. Exact expressions for the orders $\mathcal{O}(n)$ and $\mathcal{O}(n^{5/2})$ in the pressure including all orders in e^2 were derived by using cluster expansion method and Green's function methods in [10, 12-14, 35, 36] and by using different techniques in [26-28]. We will not give a complete survey about the existing exact results about the lower orders in the density but will instead direct the following study on the most important higher order terms. However let us still make a remark about the foundation of the Brillouin-Planck-Larkin partition function. There seems to be a neverending discussion about the Brillouin-Planck-Larkin partition function. Just to give one example, Starostin and Roerich [46] derived recently a quite different expression.

Our point of view which is based in particular on the results obtained here is the following:

(i) The quantum-statistical result for the contribution $\mathcal{O}(n^2)$ to the pressure is exact. The result of several groups working with independent methods give the same results (see e.g. [14, 26]).

(ii) The BPL-partition function is asymptotically in full agreement with the second virial coefficient. There is a small freedom in the special choice of a partition function, as far as the asymptotics is not violated. However so far no good reason is to be seen, why we should leave the BPL- partition functions and make another choice [24].

(iii) According to Onsager's bookkeeping theorem, in a correct theory any small change of the partition function does not matter, the final results for any physical observables are stationary with respect to small changes [24].

Some of the critics of the BPL partition function is just based on intuitive arguments and is not based on a serious revision of the quantum-statistical foundations given in the work of Vedenov and Larkin and the subsequent papers.

Admittedly the quantum statistics behind this formula is very complicated but nevertheless it is strict. With respect to the intuitive arguments we only want to say, the formal reasons for having no terms of order $O(e^0)$ and $\mathcal{O}(e^4)$ in the Planck-Brillouin-Larkin formula is that these terms were consumed already in the formulation of the ideal terms which are $\mathcal{O}(e^0)$ and the screening terms $\mathcal{O}(e^3)$, and $\mathcal{O}(e^4)$ which arise formally by summing up an infinite series in $\mathcal{O}(e^n)$. This procedure consumes some terms of order $\mathcal{O}(e^3)$, and all contributions of order $\mathcal{O}(e^4)$ to the partition function which cannot appear anymore in exact expression for the partition function, if double counting of terms is strictly avoided. Another, formally equivalent argumentation is based on the fact of compensation of the terms below the series limit and above the series limit is most essential in this context [36]. We want to underline again that there is no unique choice of the partition function for hydrogen, there are other possibilities [24]. Essential is however, that all possible choices should be compatible with the quantum statistical formulae given above. That means among the many admittable versions of partition functions, the essential criterion for correctness is Onsager's bookkeeping rule. In other words the question is: Is there a correct counting of all contributions or not. Is there a check for doublecounting or not [24]. As a matter of fact, all partition functions which work without a BPL- or equivalent subtraction of first order terms may have the problem of doublecounting of certain diagrams of order $\mathcal{O}(e^n)$.

Using the asymptotic expression for $K^*(T)$, which is asymptotically exact for $k_B T \ll I$, the pressure may be written in a different form

$$\beta p(\beta, n) = n \left[1 + \frac{1}{8\sqrt{2}} n \Lambda_e^3 + \cdots \right]$$
$$+ n - \frac{\kappa^3}{24\pi} \left[1 - \frac{3\sqrt{\pi}}{8} (\kappa \lambda) + \frac{3}{10} (\kappa \lambda)^2 + \cdots \right]$$
$$- n^2 [\Lambda^3 \sigma_{\rm BPL}(T)] [1 - \beta e^2 \kappa] + \mathcal{O}(n^3 \log n). \tag{37}$$

We combined one of the two contributions from $K^*(T)$ with the ideal electron pressure and the other one with the Debye law. Further we introduced the mean thermal

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wave length $\lambda = \lambda_{ie}$ which is an effective quantum length, in some correspondence to the Debye-Hückel parameter in the classical theory (μ - reduced mass). We have to note here again that the results given above are not fully complete in the order $n^{5/2}$ and in missing higher order corrections. However this problem seems to be not relevant for the bound state problem. A formally equivalent form of Eq. (37) which shows better the general structure is the following

$$\beta p(\beta, n)$$

$$= n \left[1 + \frac{1}{8\sqrt{2}} n \Lambda^{3} + \cdots \right] + n - n\gamma \left[(1 - \beta e^{2} \kappa + \cdots) + \cdots \right]$$

$$- \frac{\kappa^{3}}{24\pi} \left[1 - \frac{3}{2} \gamma + \cdots \right] \left[1 - \frac{3\sqrt{\pi}}{8} (\kappa \lambda) + \frac{3}{10} (\kappa \lambda)^{2} + \cdots \right]$$

$$+ \mathcal{O}(n^{3/2}), \qquad (38)$$

$$\gamma = \Lambda^3 \sigma_{\rm BPL}(T) = 8\pi^{3/2} \lambda_{ie}^3 \sigma_{\rm BPL}(T). \tag{39}$$

Summarizing we may state that there exists an asymptotically exact expression for the pressure of hydrogen in the limit of small densities, which is given in the formulae Eqs. (37), (38) which we have drawn in Fig. 2. This approach is compatible with the extended Saha theory what encourages us to proceed in this direction. Let us first investigate Eq. (38) as it is, without completing the series. Looking at the curves in Fig. 2 we see that the screening effects and the corresponding BPL effects tend to lower the pressure. We see that this lowering is too large at increasing densities.

The structure of Eq. (38) suggests that the expressions in the parenthesis may be only the first terms of some infinite series with respect to $n\Lambda^3$, κ and γ . Just to give a first example: The first series is evidently nothing else than the ideal electron pressure, which may easily be extended to an infinite series using the known expression for the ideal Fermi pressure. There is no question that all these terms appear in the full cluster series. In order to find the higher orders in the other series we have to use a more complicated procedure which is based on fugacity expansions [10, 13, 47]. According to our general philosophy, the strong decay of the pressure observed in second order density expansions (Fig. 2) is due to the fact that important higher order terms corresponding to some infinite series were omitted in the derivation based on the canonical ensemble. To circumvent this difficulty is not an easy task. We mention that Alastuey et al. [27,28] succeeded already in deriving a Saha type expression in summing up an infinite series of terms. Here we follow

a similar idea but go an alternative way through fugacity expansions. As shown much earlier, fugacity-like expansions are equivalent to mass-action laws. This was used in plasma theory by Bartsch, Ebeling and others [10, 13, 47] and worked out in large detail by Rogers et al. [48]. We mention that Rogers et al. succeeded to show that extended fugacity expansions allow an excellent description of measurements for the oscillation modes of the sun [49].

4 Combined density-fugacity expansions including nonlinear ring and ladder contributions

At higher densities a systematic quantum statistical approach to the pressure of plasmas can be given based on

the Green's function representations of the pressure. The central relation for the pressure reads [13, 14, 18]

$$p(\beta, \mu_e, \mu_i) = p_{id}$$

- $\frac{1}{2V} \int_0^1 \frac{d\lambda}{\lambda} \int d1 d\tilde{1} V(1\tilde{1}) G_2(1, \tilde{1}, 1^{++}, 1^+; \tilde{t}_1 = t_1^+), \quad (40)$

where $1 = {\mathbf{p}_1, \sigma_1}$ denotes momentum and spin variables. The Green's function representation works in the grand canonical ensemble which provides us a series in the fugacities

$$z_a = \frac{2s_a + 1}{\Lambda_a^3} \exp(\beta \mu_a) = n_a \exp(\beta \mu_a^{\text{ex}})$$

instead of a series in the densities. Note that μ_a^{ex} is the excess part of the chemical potential (the part beyond the ideal Boltzmann term) and that therefore $z_a \rightarrow n_a$ for small densities. At small fugacities (densities) the pres-

Figure 2 (online color at: www.ann-phys.org) Pressure related to the total classical pressure $2nk_BT$ at constant temperatures T = 0.1I and T = 0.15I in dependence on the density (log-scale). The result of density expansions up to 2nd order virial terms (red and green) are compared with corresponding fugacity expansions.

sure is given as an expansion in the fugacities

$$\beta p = \beta \sum_{a} z_{a} + \sum_{ab} z_{a} z_{b} b_{ab}(\kappa_{g}) + \sum_{abc} z_{a} z_{b} z_{c} b_{abc}(\kappa_{g})$$
$$+ \sum_{abcd} z_{a} z_{b} z_{c} z_{d} b_{abcd}(\kappa_{g}) + \cdots, \qquad (41)$$

where

$$\kappa_g^2 = 4\pi\beta\sum_a z_a e_a^2$$

is the grand-canonical screening length. The structure of this series is similar to the density series, we have e.g. similar screening effects, however there are some differences. e.g. the fugacity series contains more diagrams. The contribution of bound states of an electron and an ion (atom) is contained in the contribution $z_e z_i b_{ei}$, the contribution of molecules is contained in the term $z_e^2 z_i^2 b_{eiei}$.

General expressions for screened fugacity series were written down first by Montroll and Ward, Vedenov and Larkin and explicit calculations were done by DeWitt and Larkin [8, 33, 50]. Semiclassical fugacity series were given by several workers [13, 47].

We are interested here mostly in the bound state effect and may simplify our calculations by using approximations allowed for the region $T \leq 157000$ K where the bound states appear. Further in order to simplify the quite complicate evaluation of these diagrams we restrict the theory to the region of non-degenerate plasmas. As to be seen from the above formulae, the difference of the masses of electrons and ions (protons) do not play a role in the region $|\xi_{ab}| \ll 1$ and the plasma behaves asymptotically like a system with equal masses and equal relative De Broglie wave lengths $\lambda = \lambda_{ie}$. In the lowest approximation corresponding to Eq. (37) we find for low temperature hydrogen plasmas the following results valid up to order $O(z^{5/2})$ [13, 14].

$$\beta p = z_e + z_i + \frac{\kappa_g^3}{12\pi} f(\kappa_g \lambda) + 8\pi z_e z_i \lambda^3 \exp\left[1 + \beta e^2 \kappa_g\right] \sigma_{\rm BPL}(T) + \cdots, \qquad (42)$$

where the grand-canonical screening quantity is now defined by

 $\kappa_g^2 = 4\pi\beta(z_e + z_i)e^2,$

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and the quantum-statistical ring function is a transcendent function [10, 13] with the series

$$f(x) = \left[1 - \frac{3\sqrt{\pi}}{16}x + \frac{1}{10}x^2 + \cdots\right].$$
(43)

The relations between densities and fugacities are given by

$$n_{i} = z_{i} \frac{\partial \beta p}{\partial z_{i}} = z_{i} + \frac{\kappa_{g}^{2}}{16} \frac{\partial}{\partial \lambda} (\kappa_{g} \lambda f(\kappa_{g} \lambda)) + 4\pi z_{e} z_{i} [1 + \beta e^{2} \kappa_{g} + \cdots] \lambda^{3} \sigma_{\text{BPL}}(T) \cdots,$$
(44)

$$n_{e} = z_{e} \frac{\partial \beta p}{\partial z_{e}} = z_{e} + \frac{\kappa_{g}^{2}}{16} \frac{\partial}{\partial \lambda} (\kappa_{g} \lambda f(\kappa_{g} \lambda)) + 4\pi z_{e} z_{i} [1 + \beta e^{2} \kappa_{g} + \cdots] \lambda^{3} \sigma_{\text{BPL}}(T) \cdots .$$
(45)

We will show that going by iterations from the fugacity variable to the density this result occurs to be equivalent to Eq. (37). In order to proceed with this complicated system of equations we go to the special case of non-degenerate hydrogen at lower temperatures $T \leq I/k_B$ and satisfying $n\Lambda^3 \leq 1$. As shown above, in this region the differences of the electron ion masses do not matter and we my assume $z_e = z_i = z$. In this approximation, Eqs. (42), (44), (45) read

$$\beta p = 2z + \frac{\kappa_g^3}{12\pi} f(\kappa \lambda) + 8\pi z^2 \lambda^3 \sigma_{\rm BPL}(T) + \cdots, \qquad (46)$$

$$n = z + \frac{\kappa_g^2}{16\pi} \frac{\partial}{\partial \lambda} (\kappa_g \lambda f(\kappa_g \lambda)) + 8\pi z^2 \lambda^3 \sigma_{\rm BPL}(T) \cdots .$$
(47)

In order to represent the pressure by densities, we neglect, in a first step, the bound state contributions. Expressing first the fugacities by densities we find

$$z = n \exp\left[-\frac{\beta e^2 \kappa}{2} G(\kappa \lambda)\right],\tag{48}$$

where [10, 13]

$$G(x) = 1 - \frac{\sqrt{\pi}}{4}x + \frac{1}{6}x^2 - \cdots.$$
 (49)

Then excluding the fugacities from the series step by step we arrive at the representation

$$\beta p = 2n - \frac{\kappa^3}{24\pi} \phi(\kappa \lambda). \tag{50}$$

Here $\phi(x)$ is another transcendent function, related to f(x) and f'(x), the so-called pressure ring function which is an analogue of the Debye-Hückel ring function [13, 33, 50]. The first terms of the series expansion read

$$\phi(x) = \left[1 - \frac{3\sqrt{\pi}}{8}x + \frac{3}{10}x^2 - \cdots\right].$$
(51)

A full representation of this quite complicated transcendent function may be obtained by methods similar as described in [13]. We found the infinite series

$$\phi(x) = 1 - \frac{\sqrt{\pi}}{3} \sum_{k=1}^{\infty} \frac{(k+1)(k+3)x^{2k-1}}{2^k k!} + \frac{1}{3} \sum_{k=1}^{\infty} \frac{(k+1)(k+3)x^{2k}}{2^k (2k-1)!!}.$$
(52)

The asymptotic behavior of this function is given by

$$\phi(x) = \frac{3}{x^2}.\tag{53}$$

This suggests the convenient Padé representations, which will be used in the calculations

$$\phi(x) = \frac{24}{24 + 9\sqrt{\pi}x + 8x^2}; \quad G(x) = \frac{1}{1 + (\sqrt{\pi}/4)x}.$$
 (54)

A representation by error functions reads

$$\phi(x) = \frac{3\sqrt{\pi}}{3x} + \frac{4\sqrt{\pi}}{x^3} - \frac{4\sqrt{\pi}}{x^3} \exp[x^2/4] - \frac{\sqrt{\pi}}{3x} [1 - \Phi(x/2)] + {}_1F_1.$$
(55)

The saturating behavior of the nonlinear function $\Phi(x)$ in comparison to its linear approximation is shown in Fig. 3.

We summarize the results we obtained so far: By means of grand-canonical methods we succeeded to obtain some infinite series in the screening parameter κ instead of the first linear and quadratic terms only, as we found in the density series, (see Eq. (38)). This summing up a series in κ which leads to a saturating function (see Fig. 3) and improves very much the behavior at larger densities as we will show.

In our representation Eq. (38), there appears also another series expansion in the density parameter γ . In order to sum up this series in γ , we proceed in a similar way. In order to understand the structure of the series we study at first the bound state contributions, neglecting the contributions coming from the ring term and from the terms due to degeneracy. Further we replaced the factor $1 - \beta e^2 \kappa$ by a more smooth exponential function $\exp(-\beta e^2 \kappa)$. This yields the simple quadratic equation

$$n = z + 8\pi z^2 \lambda^3 \sigma_{\rm BPL}(T). \tag{56}$$

By the way, this formula shows already that the fugacities of the electrons and the protons should be rather small in the bound state region where the partition function is large. The quadratic equation for the fugacities can be

Figure 3 (online color at: www.ann-phys.org) The nonlinear ring function $\phi(x)$, which determines the contribution of screening to the pressure (blue line). Further we show the nonlinear ladder function A(x) (red line) which determines the contribution of the bound states to the pressure as well as the linear approximation, which gives only the lowest order to the second virial coefficient (thin line).

easily solved. This way we find, after elimination of the fugacity *z* (alternatively using the *A*-function):

$$z^{0} = nA(\gamma); \quad A(x) = \frac{1}{2x} [\sqrt{1+4x} - 1];$$

$$\gamma = 8\pi n\lambda^{3}\sigma_{\rm BPL}(T).$$
(57)

We note the following series expansion and large x asymptotic of the nonlinear function A(x):

$$A(x) = 1 - x + 2x^2 - \dots; \quad A(x) = \frac{1}{\sqrt{x}} + \dots.$$
 (58)

The saturating behavior of the nonlinear function A(x) in comparison to its linear approximation is shown in Fig. 3. This shows that in the region of large partition functions the fugacities disappear as

$$z^{0} = \frac{1}{\sqrt{8\pi n\lambda^{3}\sigma_{\rm BPL}(T)}}.$$
(59)

This is important for the understanding, why the fugacity series has, in the region of bound states, a better convergence than the density series. The better convergence is due to the fact that the fugacities disappear in the region where the bound state contribution is large.

Introducing the zeroth step of iteration z^0 into the pressure we find

$$\beta p^{0} = z^{0} + (z^{0} + 8\pi z^{2} \lambda^{3} \sigma_{\text{BPL}}(T))$$

= $z^{0} + n = n(1 + A(\gamma)).$ (60)

This surprising representation tells us that the Saha function A(x) appears in the pressure in a quite natural way through the fugacity expansion. We have introduced elements of the chemical picture without using the notation of chemical species explicitely. The new "semichemical representation" includes the nonlinear function $A(\gamma)$ which saturates at large densities (see Fig. 3). The new representation contains all terms in the density up to n^2 as well as several higher order terms in the density contained in the nonlinear $A(\gamma)$ which is related to the Saha function defined by Eq. (4). We need this kind of functions to reproduce an ideal Saha-type behavior. The fugacity series contains more terms than the density series, so we may expect that some of the difficulties connected with density expansions, as the strong decrease of the pressure with increasing density shown in Fig. 2 may be avoided. Indeed, a representation of the curve corresponding to Eq. (42) shows a more reasonable behavior with increasing density which is much closer to the Sahatype behavior. As the curves shown in Fig. 2 demonstrate, the pressure according to the fugacity expansion goes to saturation. This is due to the fact that the fugacity is not fixed, it decreases with increasing value of the partition function and this way limits the growth. Evidently the fugacity expansions provide a more correct description of the bound state contributions as demonstrated in Fig. 2.

We find that density as well as fugacity expansions have both advantages as well as disadvantages:

1) The density expansion describes well the screening effects but it fails to cope with the diverging contributions from the screening terms and from the BPL-partition function.

2) The fugacity expansion corresponds to an infinite density series including the partition function σ . If σ is large, then the fugacity goes to zero what guarantees even at large densities always finite contributions to the pressure. This is true for the screening contributions and for the bound state contributions; any strong increase of contributions suppresses the fugacities. In general, the fugacity expansions are more smooth.

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In conclusion we may expect that the best representation is obtained by extended density expansions which contain additional contributions corresponding to the important damping terms in the fugacity expansions. We may expect that this procedure provides Saha-like terms. This kind of mixed expansions combines the positive features of both expansions avoiding the negative features. In order to demonstrate this we started here from the density expansions and used the fugacity expansions mostly only for finding the right continuation of the infinite series with respect to the γ - parameter and the $\kappa\lambda$ parameter.

Following this line and including all terms up to known orders in the densities and fugacities, we obtain, in a first step, the following expression for the fugacities

$$z = na';$$

$$a' = A \left(8\pi n\lambda^3 \exp[-\beta e^2 \kappa G(\kappa \lambda)] \sigma_{\rm BPL}(T) \right).$$
(61)

Note that here the second term plays the role of an effective density of the bound particles. For the pressure we find, including again the full electronic Fermi pressure, the following relatively simple formula which contains the three nonlinear functions which we discussed in the foregoing part

$$\beta p = p_e^F + nA(\Gamma) - \frac{\kappa'^{3/2}}{24\pi} \phi(\kappa'\lambda), \qquad (62)$$

with the definitions of renormalized values of the Saha parameter and the screening parameter

$$\begin{split} &\Gamma = 8\pi n\lambda^3 \exp[-\beta e^2 \kappa G(\kappa'\lambda)]\sigma_{\rm BPL}(T); \\ &\kappa' = \kappa A^{1/2}(\gamma). \end{split} \tag{63}$$

This provides us with a closed and relatively simple formula for the pressure, including the Fermi pressure. The underlying assumption is that all electrons contribute to the Fermi pressure. This result for the pressure according to the combined density-fugacity representation Eq. (62) which includes all contributions up to second order in the fugacity is, for three temperatures, shown in Fig. 4.

The good convergence of the new expression including the nonlinear functions A(x) and $\phi(x)$ based on the fugacity series is explained by the fact that the fugacities of the electrons and the protons are rather small in the bound state region. We notice again that this result comes from an extension of the density expansion including all quadratic terms from the fugacity virial expansions and the full Fermi pressure of the electrons. We see that the overall behavior of the new representation is much better than that of pure density or pure fugacity representations. This way we may conclude that the most appropriate description of Coulomb systems is by density expansions and including some elements from the fugacity expansions. The physics behind is that the fugacity expansions describe well the saturable forces between bound states in Coulomb systems, but on the other hand the additive long range Coulomb forces and the screening effects are better represented by density series.

We note that the new theory based on Eqs. (62), (63) is consistent with the Saha theory and in particular also with the Saha-Debye-Hückel theory [13]. Comparing the curves based on the extended theory, Figs. 4 and 5 with Fig. 2, we see that the overall behavior of the new extended density representations is much better than that of pure density or pure fugacity representations. Further we may state that the new extended analytical theory is,

Figure 4 (online color at: www.ann-phys.org) Equation of state (EOS) calculated from an extended density representation including the nonlinear ring and ladder functions $\phi(x)$, A(x) and including the Fermi pressure of the electrons. We show the pressure related to the classical ideal pressure $2nk_BT$ at constant temperatures $T = 20\,000, 30\,000, 50\,000$ K in dependence on the density (in log scale). For the temperature $T = 20\,000$ K (lowest, green line) we give a comparison with an earlier calculated curve based on the chemical picture and and Padé approximations [19, 21, 22].

Figure 5 (online color at: www.ann-phys.org) Comparison of the analytical theory for T = 20000 K (magenta) and T = 50000 K (green) to the results of a numerical code developed earlier within the chemical picture [21, 22, 24, 51, 53] (blue: T = 20000 K, red: T = 50000 K). We show the relation of the pressure to the classical ideal pressure of the ions nk_BT . The largest deviations between the curves for 20000 K are due to the formation of H₂ molecules (deep minimum in the blue curve) which are not taken into account in the present analytical theory.

up to the density $n \simeq 10^{21} \text{ cm}^{-3}$, in quite good agreement with earlier calculations based an the chemical picture and quite complex numerical codes [19, 21, 22, 24, 51]

Note however that the agreement is based on a complete agreement in the lowest orders in the density, in higher orders several differences appear. Beside the good agreement with the Saha theory and in particular with the extended Saha-Debye-Hückel theory developed in [13] we list the following deviations:

1. The exponent $\exp(\beta I)$ in the Saha equation is replaced by the BPL-partition function in the quantum-statistical formula Eq. (62).

2. The linear first Debye-Hückel screening function linear in κ is replaced here by a nonlinear ring function $\phi(x)$ describing the quantum statistical ring sums in the grand canonical expansion in a good approximation.

3. The linear second virial coefficient is replaced by a nonlinear ladder function stemming from the representation in the grand ensemble.

We note that the grand canonical ensemble plays an essential role in our derivation, and further we note that the formula Eq. (62) corresponding to a second iteration is in some sense incomplete, e.g we would expect that in higher iterations more κ -terms are replaced and that more terms corresponding to a mass-action law will appear.

We underline that several of the terms beyond n^2 and z^2 are based on extrapolations which still need further confirmations. However in the region of low temperature $T < I/k_B$ and non-degenerate plasmas our rather simple formulae give a rather good behavior and describe well the transition from low density to the valley of bound states. We note however that several physical effects as e.g. plasma phase transitions [13, 19] are not yet described by the present approach. Evidently this effect appears only in higher order terms or after transition to some chemical picture [13, 19]. However this is not our aim here.

In order to describe the transition to full ionization in the degenerate region as well, additional effects have to be taken into account, in particular the symmetry between electrons and protons is lost. The disappearance of the bound states with increasing density was discussed in [40, 41]; for more information see [14, 36].

5 Discussion and conclusion

It is shown in this work that transitions from full to partial ionization in the temperature range 10^4-10^5 K which are due to the formation of H-atoms, are well described in the physical picture. Ring and ladder diagrams in higher order approximations as well as certain combinations of these basic diagrams are taken into account. In particular the divergence of the partition function of the Bohr atom and the state of art concerning this problem is discussed.

Formally the divergence of all partition functions within the frame of the Rutherford model is due to the infinite range of the Coulomb potential. Practically, a divergent partition function means that ionization is impossible, atoms are absolutely stable. As stated already by pioneers like Herzfeld, Fermi, Planck and others, this makes no sense and there should be physical reasons which make the atomic partition function finite. Among the physical effects leading to a cut of the partition function at some level, which were mentioned already by the pioneers, are: (i) The high levels $|E_s| < k_B T$ are not stable against collisions in a thermal system and are strongly perturbed by screening effects, this is represented by the BPL reduction.

(ii) At higher densities and moderate temperatures, the contribution of the partition function provided by the ladder contributions strongly increases, also the nonideality strongly increases. The total contribution of screening and bound states remains however finite since the fugacities of the charges become small. The nonideality contributions saturate and lead to a reduction of the pressure to about one half of the ideal pressure (see Fig. 4). (iii) At very high densities beyond 10^{23} cm⁻³ the electron degeneracy delocalizes the electrons and makes atoms obsolete. The nonideality decreases again, the plasma leaves the region of nonideality (screening and bound states).

The principal scheme how the formation of bound states depends on the density at a fixed temperature is shown in Figs. 5 and 6. In Fig. 5 a comparison of the theory given here for T = 20000 K, T = 50000 K with the numerical results obtained within an advanced chemical picture by minimization of the free energy is presented. With the chemical picture we mean here a description which starts from an expression for the free energy depending on the density of free charges, atoms and molecules. One can see that the overall agreement is quite reasonable, the deviations increase only at large densities beyond $n \approx 10^{21}$ cm⁻³. The largest deviations appear near the minimum of the relative pressure for hy-

Figure 6 Typical isotherms of the degree of ionization at dissociation of hydrogen for $T = 20\,000$ K in a wide density range based on the code developed earlier within the chemical picture [21,22,24,51]. We show the degree of ionization (upper curve) and the degree of protons bound in H₂ molecules (transient maximum) as a function of density in log-scale.

drogen at 20000 K and may be interpreted by the formation of molecules which were not taken into account in the present work. These interpretations are supported by the the last Fig. 6. Here we show the degrees of ionization and the degree of molecular bound states obtained within the chemical picture for T = 20000 K. The composition and state of the system was obtained by minimization of the free energy. The numerical results used in Figs. 5 and 6 for comparison were obtained with a code developed in some earlier work [24, 30, 40].

Looking at the pressure curve in Fig. 5 one observes as in earlier sections first a decrease corresponding to the formation of bound states (atoms and molecules). This decrease comes to saturation and we observe a valley of developed bound states. With increasing density the spatial occupation of the atoms and molecules as well as effects of electronic degeneracy come into play and the bound states are destroyed. The largest deviations between the curves for 20000 K are due to the formation of H₂ molecules (deep minimum in the blue curve) which are not taken into account in the present analytical theory and also effects of strong degeneracy. These rather complicated effects are not discussed here. Parallel to the destruction of the bound states we observe an increase of the relative pressure in the region of rather high densities $n > 10^{22}$ cm⁻³. The monotonic increase of the relative pressure may be interrupted at temperatures below 10000 K by a small wiggle. This is not an artifact. A rather flat region of the pressure in dependence on the densities is contained. At lower temperatures the wiggle may be even stronger and should be discussed in connection with the possible existence of plasma phase transitions [13,19-22,29,51]. Recent work in this area was reported in [54, 55] and found a phase transition. This is not our topic here, the purpose of the present work is to consider the region of lower densities and higher temperatures. Our aim is, to obtain similar results as shown in Fig. 5 without referring to the rather complex chemical description based on atoms and molecules and their interactions. Instead of the many assumptions contained in any chemical description, the calculations given here are based only on Rutherfords model of point charges and Coulomb interactions as well as on the exact principles of quantum statistics. No a priori knowledge about the properties of atoms and molecules is required. The price to pay is that we have to avoid the regions where these species are dominant. With respect to accuracy the rather simple formulae given here may not compete with the recent successful approaches to dense plasmas based on numerical methods [52-54]. We are convinced however, that in particular to experimentalists, analytical approaches may always be quite useful in comple-

menting numerical results. In conclusion, we first analyzed the basic results for the quantum statistics of the Rutherford model of matter, partially in the historical context starting with the work of Bohr, Herzfeld, Saha, Planck, and Brillouin. Then we discussed the essential results of quantum statistics since the 60th of the 20th century. The main new results obtained in this work are: For hydrogen plasmas at lower density the EOS within the physical picture was obtained using a combined density fugacity representation which contains the essential contributions describing Saha effects and nonideality. The good convergence of the fugacity series is explained by the fact that the fugacities of the electrons and the protons are rather small in the bound state region. The obtained new expression for the EOS is valid for nondegenerate plasmas and temperatures between 20000 K and 100000 K. Partial ionization, i.e., the appearance of atoms, is restricted to a valley in the density-temperature plane. There is a transition from below starting in the region of nondegenerate electrons and a transition from above beginning in the region of degenerate electrons. Both transitions are connected to different physical effects.

It has been shown that in the non-degenerate region, the most essential effects is the BPL-reduction, resulting in the combined action of collisions and screening by absorbing the orders in e^2 , e^4 and the compensation effects of the states below and above the series limit. The most essential new contributions given here are the nonlinear ring and ladder functions introduced into the EOS. These terms stemming essentially from the grand canonical representations are responsible for a good representation of the EOS in a wide range of densities and temperatures which essentially covers the range 20000 K < T <100000 K and $n < 10^{22}$ cm⁻³. As shown in [41] at high density the Pauli-Fock contributions are essential for the limitation of the number of energy levels and determine the transition from full ionization at very high densities to partial ionization. For a detailed discussion of this problem see, e.g., [14, 36].

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