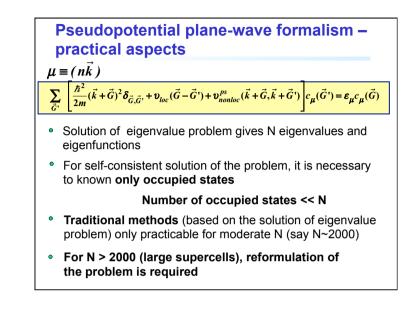
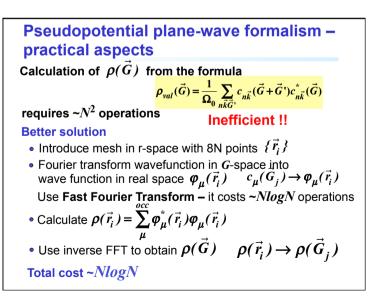
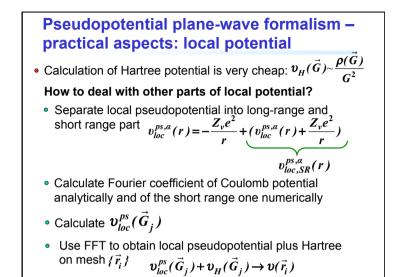
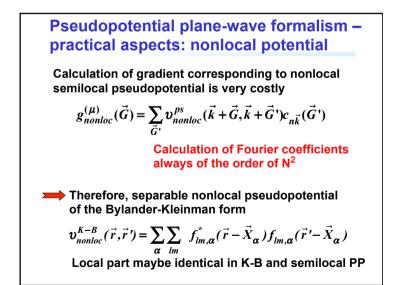


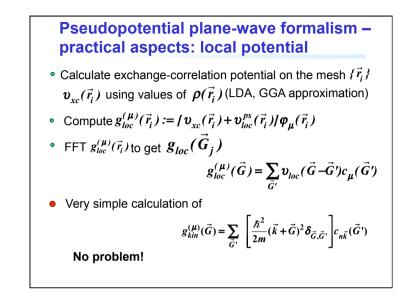
Pseudopotential plane-wave formalism – practical aspects – Iterative methods $\mu \equiv (n\vec{k}) \qquad g^{(\mu)}(\vec{G}) = \sum_{\vec{G}'} (H(\vec{G},\vec{G}') - \varepsilon_{\mu}\delta_{\vec{G},\vec{G}'})c_{\mu}(\vec{G}')$ Gradient We are looking for $\mu \ge N_{occ}$ wavefunctions • mutually orthogonal $\langle \mu | \mu' \rangle = \sum_{\vec{G}} c_{\mu}^{*}(\vec{G})c_{\mu'}(\vec{G}) = \delta_{\mu\mu'}$ • such that the gradients vanish $g^{(\mu)}(\vec{G}) = 0$ Searching procedure ? e.g. steepest descent $c_{\mu}^{(n+1)} = c_{\mu}^{(n)} + \eta g^{(\mu)}(n)$ conjugate gradient Davidson method Required: effective method to calculate gradient











Pseudopotential plane-wave formalism – practical aspects: nonlocal potential $v_{nonloc}^{K-B}(\vec{r},\vec{r}') = \sum_{\alpha} \sum_{lm} f_{lm,\alpha}^*(\vec{r} - \vec{X}_{\alpha}) f_{lm,\alpha}(\vec{r}' - \vec{X}_{\alpha})$ The knowledge of $f_{lm,\alpha}(\vec{G})$ is sufficient to calculate $g_{nonloc}^{(\alpha)}(\vec{G})$ Computational cost proportional to N, but with rather large prefactor Pseudopotential plane-wave formalism – practical aspects norm conserving PPs

- Computational schemes with norm conserving PPs
- Computational burden ~ $N \log N$
- Good transferability of the PPs
- Atoms from the first row of Periodic Table require kinetic energy cutoff of ~60-70 Ry
- **Too many plane-waves required in many applications** atoms from the first row of Periodic Table semicore d-states Even one atom of this type requires large cutoff
- → Ultra soft pseudoptentials

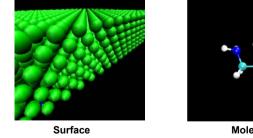
Features of the Pseudopotential Method

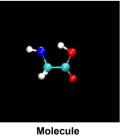
- Pseudopotential is approximation to all-electron case, but... <u>Very accurate</u>
 Comparable accuracy to AE in most cases
 Simpler formalism
 - Low computational cost •Perform calculations on 'real-life' materials

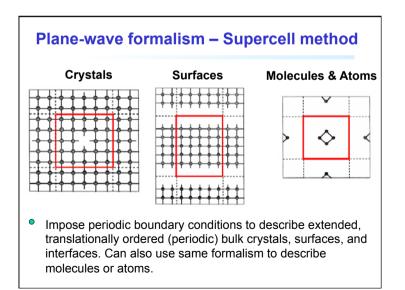
Allows full advantage of plane-wave basis sets •Systematic convergence •Easy to perform dynamics

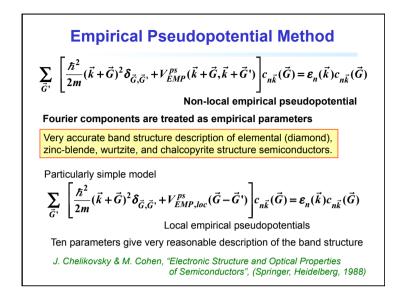
Plane-wave formalism for systems with reduced periodicity

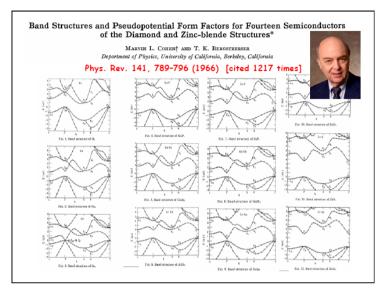
How we solve the Kohn-Sham problem when the systems are not periodic: surfaces, interfaces, clusters, molecules... and still take advantage of the plane wave formulation of the problem



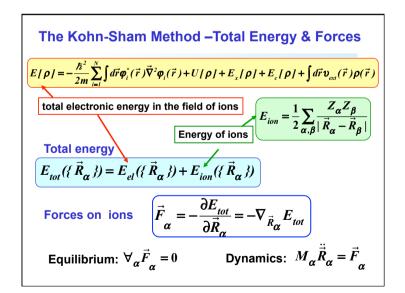


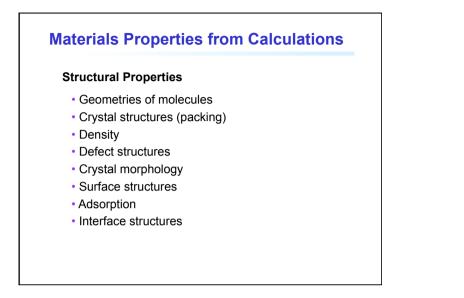


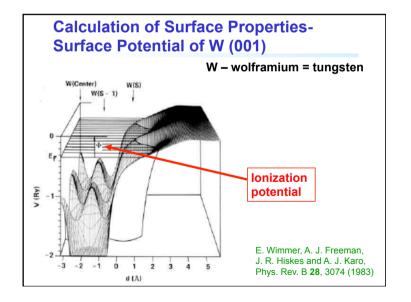


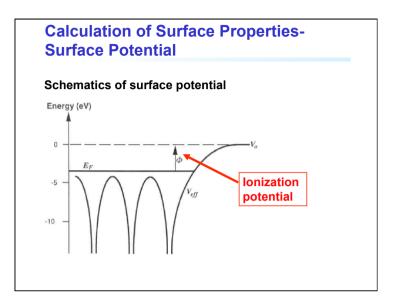


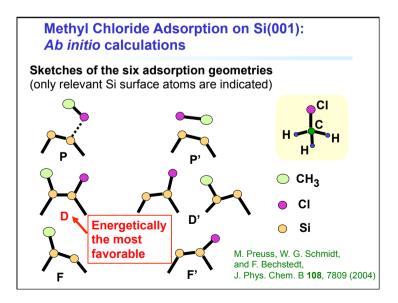


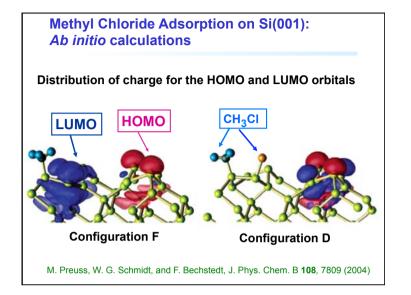












Materials Properties from Calculations (3)

Electronic, optical, and magnetic properties

- Electron density distribution electrical moments
- Polarizabilities
- · Ionization energies and electron affinities
- · Electrostatic potential, work function
- Energy band structure metal, semiconductor, insulator, superconductor
- Band offsets at hetero-junctions
- Optical spectra
- Spin density distribution, magnetic moments, crystalline magnetic anisotropy
- Magneto-optical properties (Kerr rotation)
- NMR chemical shifts
- Dielectric response
- Luminescence

Materials Properties from Calculations (2) Mechanical Properties • Compressibility • Elastic moduli • Thermal expansion coefficients

- Vibrational properties
- Hardness
- Fracture toughness

Thermodynamic Properties

- Binding energies
- Pressure induced phase transitions

Materials Properties from Calculations ?

Thermodynamic Properties

- Temperature induced phase transitions
- · Phase diagrams
 - (liquid-gas, liquid-liquid, liquid-solid, solid-solid)

Transport Properties

- Thermal conductivity
- Viscosity
- Diffusion constants

Require dynamic treatment of ion movement

⇒ Molecular Dynamics

Chemical Properties – Molecular Dynamics needed

Chemical and other properties

- Chemical reaction rates (catalytic properties, corrosion, electrochemistry)
- Reactivity with surfaces
- Photochemical properties

Molecular Dynamics – What is it ?

• Classical dynamics (given by Newton equations) of atoms in the system

$$M_{I}\vec{\vec{R}}_{I} = \vec{F}_{I}$$

$$\vec{F}_{I} = -\nabla_{I}V_{eff}(\{\vec{R}_{I}\})$$

- Goal to determine classical trajectories of all atoms in the system $\{R_I(t)\}$
- At the very heart of any molecular dynamics scheme is the question of how to describe that is in practice how to approximate the *interatomic interactions*.
- Where to get forces on atoms from ?

Molecular Dynamics

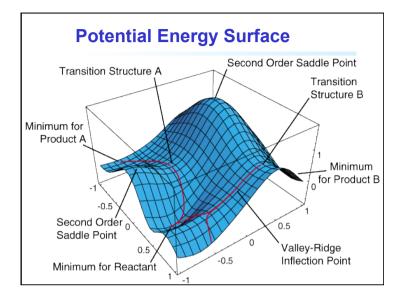
- ✓ Classical vs. *ab initio*
- ✓ Ehrenfest vs. Born Oppenheimer
- ✓ Car-Parrinello MD

Classical vs. Ab initio Molecular Dynamics

$$M_{I}\ddot{\vec{R}}_{I} = -\nabla_{I}V_{eff}(\{\vec{R}_{I}(t)\})$$

Potential energy surface

- Classical based on predefined potentials
 - either obtained from empirical data, or
 - obtained from independent electronic structure calculations
- Ab initio based on fully quantum mechanical calculations



Classical Molecular Dynamics

• Very often, the interactions can faithfully be described by additive two-body terms

$$V_{\rm e}^{\rm approx}(\{\mathbf{R}_I\}) \approx \sum_{I < J}^N v_2(|\mathbf{R}_I - \mathbf{R}_J|)$$

For example, Argon in liquid phase

E. Ermakova, J. Solca, H. Huber, and D. Marx, Chem. Phys. Lett. **246**, 204 (1995).

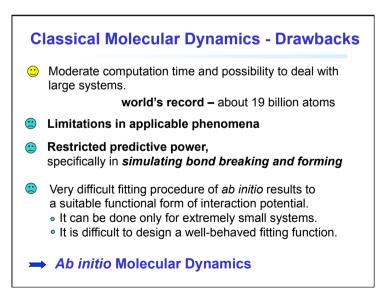
- Well established tool to investigate many-body condensed matter systems
- A lot of monographs, e.g.,
 - M. P. Allen and D. J. Tildesley, *Computer Simulation of Liquids* (Clarendon Press, Oxford, 1987; reprinted 1990).
 - D. Frenkel and B. Smit, Understanding Molecular Simulation From Algorithms to Applications (Academic Press, San Diego, 1996).

Classical Molecular Dynamics

 Typically, the full *interaction potential* is broken up into two-body, three-body and many-body contributions, long-range and short-range terms etc., which have to be represented by suitable functional forms.

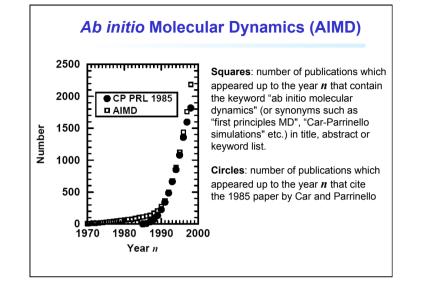
$$\begin{split} V_{\mathrm{e}}^{\mathrm{E}} &\approx V_{\mathrm{e}}^{\mathrm{approx}}(\{\mathbf{R}_{I}\}) = \sum_{I=1}^{N} v_{1}(\mathbf{R}_{I}) + \sum_{I < J}^{N} v_{2}(\mathbf{R}_{I}, \mathbf{R}_{J}) \\ &+ \sum_{I < J < K}^{N} v_{3}(\mathbf{R}_{I}, \mathbf{R}_{J}, \mathbf{R}_{K}) + \cdots \end{split}$$

• The electronic degrees of freedom are replaced by interaction potentials v_1 , v_2 , etc. and are not featured as explicit degrees of freedom in the equations of motion.





- The fitting step can be bypassed and the dynamics performed directly by calculating the inter-atomic forces (obtained from the electronic structure calculated on-the-fly) at each time-step of an MD simulation
- The methods are widely applicable and possess potential to predict new phenomena and novel materials.
 - → New Area in Materials Science Materials Design
- The methods may cost huge computation time, however
 - Progress in Computational Techniques (new solvers, new optimization techniques, etc.)
 - Progress in high-performance computers



Ab initio Molecular Dynamics (AIMD) - Theory

$$\frac{d\mathbf{P}_{I}}{dt} = -\nabla_{I} \int d\mathbf{r} \ \Psi^{\star} \mathcal{H}_{e} \Psi \quad \text{or}$$

$$M_{I} \ddot{\mathbf{R}}_{I}(t) = -\nabla_{I} \int d\mathbf{r} \ \Psi^{\star} \mathcal{H}_{e} \Psi$$

$$= -\nabla_{I} V_{e}^{\mathrm{E}}(\{\mathbf{R}_{I}(t)\})$$
• Thus, the nuclei move according to classical mechanics in an effective potential V_{e}^{E} due to the electrons.

• This potential is a function of only the nuclear positions at time *t* as a result of averaging H_e over the electronic degrees of freedom, i.e. computing its quantum expectation value $\langle \Psi | \hat{H}_e | \Psi \rangle$, while keeping the nuclear positions fixed at their instantaneous values $\{\vec{R}_I(t)\}$.

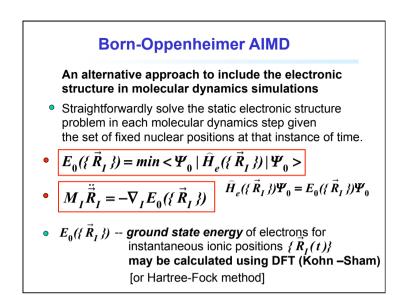
1	Fime-dependent wave equation for the electrons
•	$i\hbar \frac{\partial \Psi}{\partial t} = -\sum_{i} \frac{\hbar^2}{2m_{\rm e}} \nabla_i^2 \Psi + V_{\rm n-e}(\{\mathbf{r}_i\}, \{\mathbf{R}_I(t)\}) \Psi$
•	$= \mathcal{H}_{e}(\{\mathbf{r}_{i}\}, \{\mathbf{R}_{I}(t)\}) \ \Psi(\{\mathbf{r}_{i}\}, \{\mathbf{R}_{I}\}; t)$ Hamiltonian and wave-function are parametrically dependent on the classical nuclear positions $\{\vec{R}_{I}(t)\}$
•	Procedure of solving simultaneously Eqs. and is very often called <i>"Ehrenfest molecular dynamics"</i> .
•	It was never in widespread use for systems with many active degrees of freedom typical for condensed matter problems

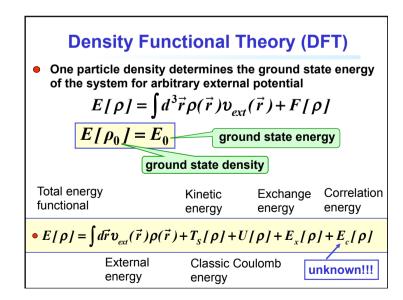
Ehrenfest molecular dynamics

In practical calculations only ground state considered

$$M_{I} \ddot{\mathbf{R}}_{I}(t) = -\nabla_{I} \left\langle \Psi_{0} \left| \mathcal{H}_{e} \right| \Psi_{0} \right\rangle$$
$$i\hbar \frac{\partial \Psi_{0}}{\partial t} = \mathcal{H}_{e} \Psi_{0}$$

- Electronic Hamiltonian is time-dependent via the nuclear coordinates
- The propagation of the wavefunction is unitary, i.e. the wavefunction preserves its norm and the set of orbitals used to build up the wavefunction will stay orthonormal





Born-Oppenheimer AIMD (cnt.)

- The electronic structure part is reduced to solving a *time-independent quantum problem*
- The time-dependence of the electronic structure is a consequence of nuclear motion, and not intrinsic as in Ehrenfest molecular dynamics.



- In Ehrenfest dynamics the time scale is dictated by the intrinsic dynamics of the electrons.
- Since electronic motion is much faster than nuclear motion, the largest possible time step is that which allows to integrate the electronic equations of motion.
- There is no electron dynamics whatsoever involved in solving the Born-Oppenheimer dynamics, i.e. they can be integrated on the *time scale given by nuclear motion*.
- However, this means that the electronic structure problem has to be solved self-consistently at each molecular dynamics step.

Car- Parrinello Method - Motivation

R. Car and M. Parrinello, Phys. Rev. Lett 55, 2471 (1985)

- A non-obvious approach to cut down the computational expenses of molecular dynamics
- It can be seen as an attempt to combine the advantages of both Ehrenfest and Born-Oppenheimer molecular dynamics.
- From an algorithmic point of view the main task achieved in ground-state Ehrenfest dynamics is simply to keep the wavefunction automatically minimized as the nuclei are propagated.
- This, however, might be achieved -- in principle by another sort of deterministic dynamics than first-order Schrödinger dynamics.

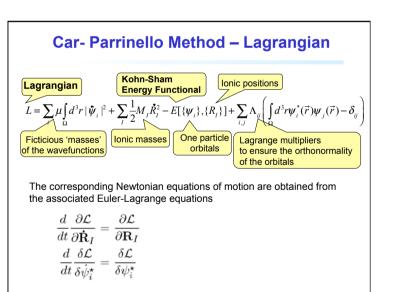
Car- Parrinello Method - Motivation

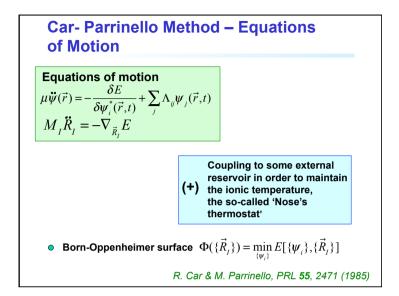
The "Best of all Worlds Method" should

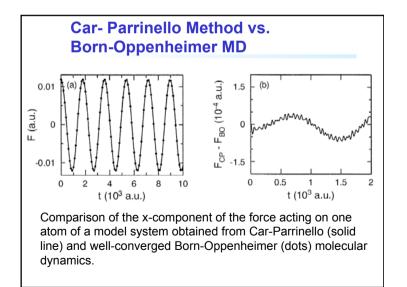
- (i) integrate the equations of motion on the (long) time scale set by the nuclear motion but nevertheless
- (ii) take intrinsically advantage of the smooth time-evolution of the dynamically evolving electronic subsystem as much as possible.

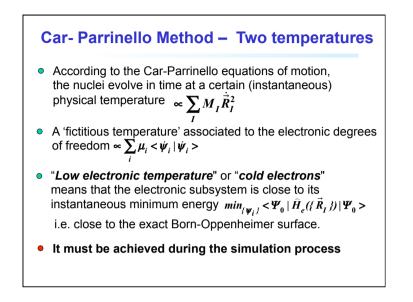
The second point allows to circumvent explicit diagonalization or minimization to solve the electronic structure problem for the next molecular dynamics step.

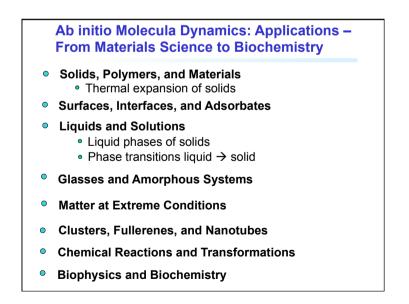
• **Car-Parrinello molecular dynamics** is an efficient method to satisfy requirement (ii) in a numerically stable fashion and makes an acceptable compromise concerning the length of the time step (i).



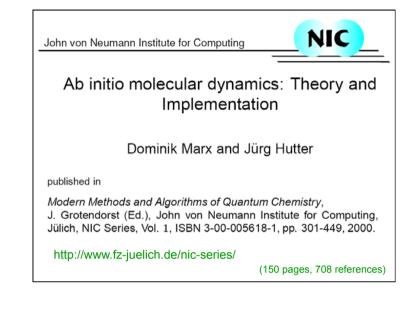












M. Bernasconi, Parrinello, Max	sion 3.3: I. Hutter, A. Alavi, T. Deutsch, St. Goedecker, D. Marx, M. Tuckerman, and M. -Planck-Institut für Festkörperforschung and IBM .h Laboratory (1995-1999).
CASTEP	Accelrys
CP-PAW	P. E. Blöchl
fhi98md	Fritz-Haber-Institut, Berlin
NWChem	developed and distributed by Pacic Northwest National Laboratory, USA.
VASP	University of Vienna

